

Dynamic Protection of Amines using 18-Crown-6

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Summary The regioselectivity of diamine monoacylation has been controlled by selective complexation with 18-crown-6 and a proton source

RECENTLY we reported a convenient method for the selective acylation of secondary amines in the presence of primary amines¹ 18-Crown-6 forms complexes with alkylammonium salts *via* three hydrogen bonds and pole-dipole interactions in the 2.7 Å cavity² We expected that rapid selective complexation of one (or more) ammonium function(s) in a polyamine substrate should permit selective functionalisation of uncomplexed sites Since dialkylammonium salts form less stable complexes owing to a reduction in hydrogen bonding, selective acylation of a secondary amino function in the presence of a primary is possible

selective monobenzoylation of ethylenediamine and homologues was also improved in the presence of 18-crown-6

The decrease in stability of crown-primary alkylammonium salt complexes with increasing steric congestion³ should permit the selective acylation of a hindered primary amine in the presence of a non-hindered function Such selection is relevant to aminoglycoside chemistry As model systems, competition in the acylation and toluene-4-sulphonylation of mixtures of benzylamine and benzhydrylamine or 3 α -(axial) and 3 β -(equatorial) amino-5 α -cholestanes⁴ were studied (Table 2) Without crown ether the less hindered (benzyl- or 3 β - respectively) amine was principally functionalised In the presence of 18-crown-6 the ratio of hindered non hindered amides was increased Consistent with sterically selective complexation³ dicyclohexyl 18-crown-6 (entries 10, 11) was superior to 18-crown-6 In the

TABLE 1 Selective acylation of diamines RNH[CH₂]_nNH₂^a

Entry	Equiv of 18-crown-6	Percentage yields of products			
		RN(Ts)[CH ₂] _n -NHTs	RN(COAr)[CH ₂] _n -NHTs	RN(Ts)[CH ₂] _n -NHCOAr	RN(COAr)[CH ₂] _n -NHCOAr
1	0 ^b	30	16	0.2	37
2	1	12	63	traces	4
3	2	traces	79	traces	15
4	0 ^b	40	1	4	42
5	1	12	51	1	11
6	2	6	61	0.5	5
7	2	0	69	0	22
8	0 ^b	43	8	—	27
9	1	20	56	—	17
10	2	4	76	—	3
11	0 ^b	40	6	—	34
12	1	11	41	—	26
13	2	5	79	—	10
14	0 ^b	42	1	—	47
15	1	18	49	—	23
16	2	5	64	—	24
17	0 ^b	37	0	—	45
18	1 ^b	15	31	—	29
19	2 ^b	9	32	—	32

^a R = Me (entries 1—7) and H (8—19) Ar = Ph (1—3, 7—19) and C₆H₄-4-NO₂ (4—6), *n* = 2 (1—6, 8—10), 3 (7, 11—13), 4 (14—16) and 8 (17—19) Typically benzoyl chloride and triethylamine were added in sequence to *N*-methylethylenediammonium di(toluene-4-sulphonate) and 18-crown 6 (1 mmol each) in dichloromethane (10 ml) When reactions were complete (t.l.c.) toluene-4-sulphonyl chloride (1 mmol), triethylamine (4 mmol), and an excess of potassium chloride were added Yields refer to pure compounds isolated by direct chromatography on Merck Kieselgel H ^b Heterogeneous reactions It must be assumed that the high yield of RN-(COAr)[CH₂]_nNHCOAr in the blank reactions followed in part from the low solubility of the RNH₃⁺[CH₂]_nNH₃⁺2TsO⁻ salts However, the increase in the yield of RN(COAr)[CH₂]_nNH₂s with increase in crown ether from 1 to 2 equiv is consistent only with selective complexation

Herein we report dramatic improvements in diamine monoacylation using dynamic protection (Table 1) For example the reaction of *N*-methylethylenediamine with benzoyl and toluene-4-sulphonyl chlorides in sequence† gave *N*-benzoyl-*N*-methyl-*N*'-toluene-4-sulphonylethylenediamine (16%) In the presence of 18-crown-6 the yield was increased to 79% Surprisingly (entries 8—19) the

steroid examples exclusive axial substitution was observed in the presence of *N*-benzylmono-aza-18-crown-6⁵

The advantage of the aza-crown was emphasised by competition experiments between benzylamine and *N*-benzyl-iso-propylamine Since the rate of tosylation of the latter was slow, a 62% yield of *N*-benzyl-*N*-isopropyl-toluene-4-sulphonamide was only obtained when the

† The diamine was used as its di-toluene-4-sulphonate and, to facilitate chromatographic separation, after acylation remaining amino functions were toluene-4-sulphonylated

TABLE 2. Selective acylation and sulphonylation of amines^a

Entry	Equiv of crown ether	Amine	Ammonium salt	% Amides ^b	Hindered amide ^b mol fraction
1	0 ^c	PhCH ₂ NH ₂	Ph ₂ CHNH ₃ ⁺ TsO ⁻	96	0.61
2	1	"	"	85	0.78
3	2	"	"	79	1.00
4	0 ^c	"	"	93	0.31
5	1	"	"	92	0.47
6	2	"	"	95	0.52
7	0 ^c	"	"	95	0.04
8	1	"	"	97	0.30
9	2	"	"	91	0.44
10	1	"	"	98	0.59
11	2	"	"	98	0.71
12	0 ^c	PhCH ₂ NH ₂	PhCH ₂ Pr ⁺ NH ₂ +TsO ⁻	100	≤ 0.02
13	1	"	"	98	0.31
14	2	"	"	99	0.46
15	2	"	"	96	0.60
16	1	"	"	97	0.55
17	2	"	"	97	0.62
18	2	"	"	95	0.65
19	0	3β- : 3α-Amino-5α-cholestanes	CF ₃ CO ₂ H 1 : 1 : 1	83	0.12
20	1	"	"	85	0.40
21	2	"	"	81	0.59
22	0	"	"	96	0.26
23	1	"	"	88	0.47
24	2	"	"	88	0.70
25	1	"	"	84	1.0
26	2	"	"	87	1.0

^a Reactions were carried out using 18-crown-6 (entries 2, 3, 5, 6, 8, 9, 13–15, 20, 21, 23, and 24), dicyclohexyl-18-crown-6 (Fluka AG) (10, 11), and *N*-benzylmono-aza-18-crown-6 (16–18, 25, 26) with (CF₃CO)₂O (1–3), PhCOCl (4–6), TsCl (7–18, 22–26), or Ac₂O (19–21) as electrophile. ^b The ratios of amides were determined by n.m.r. spectroscopy (± 0.02) (entries 12–18); all other ratios refer to pure isolated compounds. Typically toluene-4-sulphonyl chloride and then, over 5 min, triethylamine (1 mmol each) were added to a solution prepared from 18-crown-6, benzylamine, and benzhydrylammonium toluene-4-sulphonate (1 mmol each) in dichloromethane (10 ml) [or (entries 19–26) from 3α- and 3β-amino-5α-cholestanes and CF₃CO₂H (1:1:1)]. Chromatography on Merck Kieselgel H gave *N*-benzyl (0.68 mmol) and *N*-benzhydryl- (0.29 mmol) toluene-4-sulphonamides. In entries 15 and 18 the triethylamine was added over 1 week. ^c Heterogeneous reactions.

triethylamine was added slowly (1 week rather than 5 min) after the toluene-4-sulphonyl chloride (Table 2, entry 18).

Clearly dynamic protection provides a more convenient simple alternative to classical protection group methodologies.

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